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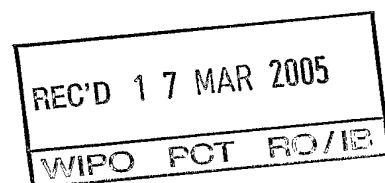
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Koninklijke Philips Electronics N.V.  
Groenewoudseweg 1  
5621 BA Eindhoven  
PAYS-BAS

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Lamp

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## Lamp

The present invention relates to a lamp, in particular a metal halide lamp, comprising a discharge vessel surrounded by an outer envelope with clearance and having a ceramic wall which encloses a discharge space filled with a filling comprising an inert gas, such as xenon (Xe), and an ionizable salt, wherein in said discharge space two electrodes are  
5 arranged whose tips have a mutual interspacing so as to define a discharge path between them.

In this description and these claims the ceramic wall is understood to mean both a wall of metal oxide such as, for example, sapphire or densely sintered polycrystalline  $\text{Al}_2\text{O}_3$  and metal nitride, for example,  $\text{AlN}$ . According to the state of the art these ceramics  
10 are well suited to form translucent discharge vessel walls.

Such a lamp is generally known. Both electrodes are each supported by a current conductor entering the discharge vessel. The current conductors consist of a first part  
15 made of an halide resistant material, such as a  $\text{Mo-Al}_2\text{O}_3$  cermet, and a second part made of niobium. Niobium is chosen because this material has a coefficient of thermal expansion corresponding to that of the discharge vessel in order to prevent leakage of the headlamp.

Disadvantages of the known lamp are the following. A central part of the discharge vessel thereof has on both sides narrow end parts or extended plugs (i.e. elongated  
20 end parts) that are connected by way of sintering to the central part of the discharge vessel and that enclose the current conductors. However, as said plugs are remote from the discharge path, they function as cooling fins, so that part of the lamp filling (i.e. salts) may condense in a void between each current conductor and the (wall of the) extended plugs. Said condensation may lead to color instability of the headlamp. De-mixing of salt components  
25 generally leads to color instabilities (for example, if the filling contains NaCe-iodide, more Na than Ce will creep into said voids). In order to obtain a light efficacy as high as possible, preferably rare earth metal iodides as  $\text{CeI}_3$ ,  $\text{PrI}_3$ ,  $\text{LuI}_3$  and/or  $\text{NdI}_3$  are added to the filling. However, these salts (especially if larger mole fractions are applied) are aggressive and will attack the ceramic wall of the discharge vessel. Further, said wall attack -close to the

discharge path- may lead to scattering/absorbing of light with all negative consequences involved for the light distribution. Finally, the light output as function of time should be as stable as possible. If salt reacts with other lamp parts and thus disappears, for example, said light output (and thus maintenance) will drop.

5 From WO 99/53522 and WO 99/53523 metal halide lamps are known which posses an improved lumen maintenance due to the existence of a W-halide cycle during lamp operation. The W-halide cycle which itself is of very complex nature and for which the presence of Ca in the filling is imperative, causes that tungsten evaporated from the hot tips of the electrodes is deposited back on parts of the electrodes being somewhat cooler, instead  
10 of deposition on the wall of the discharge vessel. Thus the W-halide cycle counteracts wall blackening. The known lamps have however a relative modest lumen output.

It is an object of the invention to obviate these disadvantages, particularly to  
15 propose a metal halide lamp operating in such a way that said corrosion of the (wall of the) extended plugs and said color instability is avoided.

In order to accomplish that objective a lamp of the type referred to in the introduction according to the invention is characterized in that said ionizable salt comprises NaI, TII,  $\text{CaI}_2$  and X-iodide, wherein X is one or more elements selected from the group  
20 comprising rare earth metals. Thus X can be formed by a single element or by a mixture of two or more elements. Preferably, X is selected from the group comprising Sc, Y, La, Ce, Pr, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Nd. More preferably, X is selected from the group comprising Ce, Pr, Nd, that is cerium, praseodymium and neodymium. Extensive research has surprisingly shown that salts comprising NaI, TII,  $\text{CaI}_2$  and X-iodide are non-aggressive  
25 and only slightly sensitive for large variations in lamp power and thus in coldest spot temperature, for example at the location of the voids mentioned above, and these salts exhibit relatively less tendency to segregation, i.e. changes in salt mix ratio at the coldest spot due to for instance corrosion or transport of said salts, and thus making the lamp relatively insensitive for performing color shifts due to segregation. For completeness' sake it is noted  
30 that Na, Tl, Ca and I stand for natrium, thallium, calcium and iodine, respectively.

In a preferred embodiment of a lamp in accordance with the invention X being the total amount of rare earth, the molar percentage ratio X-iodide/(NaI + TII +  $\text{CaI}_2$  + X-iodide) lies above 0% up to maximum 10%, in particular between 0,5 and 7%, more in particular between 1 and 6. For a too low amount of X experiments have learned that the

electrodes reach too high values of temperature to operate satisfactory. With amounts of X above the indicated maximum it turns out that it is impossible to maintain a W-halide cycles in the discharge vessel during lamp operation.

Preferably, X being the total amount of rare earth, the molar percentage ratio  $\text{CaI}_2/(\text{NaI} + \text{TII} + \text{CaI}_2 + \text{X-iodide})$  lies between 20 and 95%. When the amount of  $\text{CaI}_2$  is chosen outside the indicated range the W-halide cycles will not properly develop in the discharge vessel during lamp operation.

In another preferred embodiment of a lamp according to the invention the amount of NaI, TII,  $\text{CaI}_2$  and X-iodide lies between 0,001 and 0,5 g/cm<sup>3</sup>, in particular between 0,025 and 0,3 g/cm<sup>3</sup>. The volume of the discharge vessel particularly ranges between 0,008 and 2.5cm<sup>3</sup>.

In a preferred embodiment of a lamp in accordance with the invention the filling comprises mercury (Hg). In an alternative, the lamp filling is mercury-free.

To have a lamp which during its stable nominal operation emits light having a color temperature  $T_c$  above 3500K the filling of a preferred embodiment of the lamp according to the invention also comprises a halide selected from Mn and Ir. Experiments have learnt that with the addition of a halide of Mn and Ir the color point in the color triangle having X,Y coordinates, of the light emitted by the lamp can be adjusted primarily along the X-axis of the color triangle. Varying of the amount of Tl halide in the filling has a major impact on adjustment along the Y-axis. Stable nominal operation means in this respect that the lamp is operated at a power and voltage for which it is designed. The designed power of the lamp is called the nominal power.

The invention also relates to a metal halide lamp to be used in a vehicle headlamp according to the invention.

Finally, the invention refers to a method for manufacturing a lamp in accordance with the invention, wherein the lamp comprising a discharge vessel surrounded by an outer envelope with clearance and having a ceramic wall which encloses a discharge space filled with a filling comprising an inert gas, such as xenon (Xe), and an ionizable salt, wherein in said discharge space two electrodes are arranged whose tips have a mutual interspacing so as to define a discharge path between them, characterized in that said ionizable salt comprises NaI, TII,  $\text{CaI}_2$  and X-iodide, wherein X is selected from the group comprising rare earth metals.

The invention will now be explained in more detail with reference to Figs. illustrated in a drawing, wherein

Fig. 1 shows a preferred embodiment of a lamp according to the invention in a side elevation;

5 Fig. 2 shows the discharge vessel of the lamp of Fig. 1 in detail, and

Fig. 3. shows a further preferred embodiment having a shaped discharge vessel.

10 Figure 1 shows a metal halide lamp provided with a discharge vessel 3 having a ceramic wall which encloses a discharge space 11 containing an ionizable filling. Two tungsten electrodes 4, 5 whose tips 4b, 5b are at a mutual distance EA are arranged in the discharge space, and the discharge vessel has an internal diameter Di at least over the distance EA. The discharge vessel is closed at one side by means of a ceramic protruding  
15 plug 34, 35 which encloses a current lead-through conductor (figure 2: 40,41,50,51) to an electrode 4,5 positioned in the discharge vessel with a narrow intervening space and is connected to this conductor in a gas tight manner by means of a melting-ceramic joint (Fig. 2: 10) at an end remote from the discharge space. The discharge vessel is surrounded by an outer bulb 1 which is provided with a lamp cap 2 at one end. A discharge will extend  
20 between the electrodes 4,5 when the lamp is operating. The electrode 4 is connected to a first electrical contact forming part of the lamp cap 2 via a current conductor 8. The electrode 5 is connected to a second electrical contact forming part of the lamp cap 2 via a current conductor 9. The discharge vessel, shown in more detail in figure 2 (not true to scale), has a ceramic wall and is formed from a cylindrical part with an internal diameter Di which is  
25 bounded at either end by a respective ceramic protruding plug 34,35 which is fastened in a gas tight manner in the cylindrical part by means of a sintered joint S. The ceramic protruding plugs 34,35 each narrowly enclose a current lead-through conductor 40,41,50,51 of a relevant electrode 4,5 having a tip 4b, 5b. The current lead-through conductor is connected to the ceramic protruding plug 34,35 in a gas tight manner by means of a melting-  
30 ceramic joint 10 at the side remote from the discharge space. The electrode tips 4b, 5b are arranged at a mutual distance EA. The current lead-through conductors each comprise a halide-resistant portion 41,51, for example in the form of a Mo--Al<sub>2</sub>O<sub>3</sub> cermet and a portion 40,50 which is fastened to a respective end plug 34,35 in a gas tight manner by means of the melting-ceramic joint 10. The melting-ceramic joint extends over some distance, for example



approximately 1 mm, over the Mo cermet 40,41. It is possible for the parts 41,51 to be formed in an alternative manner instead of from a Mo--Al<sub>2</sub>O<sub>3</sub> cermet. Other possible constructions are known, for example, from EP 0 587 238. A particularly suitable construction was found to be a halide-resistant material. The parts 40,50 are made from a metal whose coefficient of expansion corresponds very well to that of the end plugs. Nb, for example, is for this purpose a highly suitable material. The parts 40,50 are connected to the current conductors 8,9 in a manner not shown in any detail. Each of the electrodes 4,5 comprises an electrode rod 4a,5a which is provided with a tip 4b,5b.

In figure 3 (not to scale) a further preferred embodiment of the lamp according to the invention is shown. Lamp parts corresponding with those shown in figure 1 and 2 have been provided with the same reference numerals. The discharge vessel 3 has a shaped wall 2 enclosing the discharge space 11. In the shown case the shaped wall forms an ellipsoid. Alternatively, other shapes like for instance pheroid is equally possible.

In a practical realization of the lamp as represented in the drawing a number of lamps were manufactured with a rated power of 30W each. The lamps are for use as headlamps for a motor vehicle. The ionizable filling of the discharge vessel 3 of each individual lamp comprises 100 mg/cm<sup>3</sup> iodide, comprising NaI, TlI, CaI<sub>2</sub> and CeI<sub>3</sub>. The filling further comprises Xe with a filling pressure at room temperature of 16 bar. The distance EA between the electrode tips 4a,5a is 4 mm, the internal diameter Di is 1.3 mm, so that the ration EA/Di=3.1. The wall thickness of the discharge vessel 3 is 0.4 mm.

A large number of lamp embodiments according to the invention have been made and tested. In a first series lamps have been tested having a cylindrical discharge vessel with an internal diameter Di of 4mm and with a filling comprising besides mercury and xenon 71.4 mol % NaI, 2.4 mol % TlI, 23.6 mol %CaI<sub>2</sub> and 2.7 mol % CeI<sub>3</sub>. Lamp properties and test results are listed below.

Table I

Lamp no.	Nominal Power (W)	EA (mm)	Luminous efficacy (lm/W)	Color temperature T <sub>c</sub> (K)	Lifetime (h)	Lumen Maintenance (%)
1	72	18	124	2900		
2	72	18	121	2900	5000	91
3	100	18	121	2900	1000	99
4	72	14	112	3000	500	99.5

5	72	15	118	2800		
6	72	17	113	2900		
7	100	17	122	3100	3000	96
8	110	17	131	3000		
9	152	23	129	3100	1000	98

The values in the columns titled "Luminous efficacy" and "Color temperature  $T_c$ " concern the results after the lamp had been operated for 100 hours. The lumen maintenance in % stated in the last column is related to the stated lifetime in the column "Life time".

From the results shown in the Table I it is clear that the invention results in a lamp with a long and stable light output. During the life time of the lamps there occurred no significant change in the color properties of the emitted light.

In Table II main data of a further series of embodiments are given.

Table II

Lamp no.	Nominal power (W)	Internal diameter Di (mm)	Salt mix Na/Tl/Ca/Ce-iodide (mol %)	Luminous efficacy (lm/W) at 100 h	Color temperature $T_c$ (K) at 100 h	Life time (h)	Maintenance (%)
10	100.6	6.85	71/2.5/23.5/3	99.1	2953	3000	96.3
11	71.8	5.6	71/2.5/23.5/3	101.6	3081	3000	104.4
12	71.6	6.85	68.7/2.8/27.6/1	99.9	3038	5000	97.3
13	71.5	6.85	74.1/2.2/22.2/3.3	101.2	3386	5000	93.4

For the lamps nr. 10 to 13 the electrode distance EA is 7mm. Over the life time as listed of the lamps in Table II they did not display any significant change in the color properties of the emitted light.

Also a number of high wattage lamps have been made and tested. These lamps had a nominal power of 400W and were provided with a cylindrical discharge vessel. The main data are listed in Table III.

Table III

Lamp no	Di (mm)	EA (mm)	Salt mix iodide (mol %)	Luminous efficacy (lm/W) at 100 h	Color temperature T <sub>c</sub> (K) at 100 h	Life time (h)	Maintenance (%)
14	12	15	Na/Tl/Ca/Ce 48/3/48/1	112	3000	5000	90
15	12	12	Na/Tl/Ca/Ce 4/3/92/1	101	4000	5000	85
16	10	28	Na/Tl/Ca/Mn /Ce 35/3/35/25/1	96	4100	500	99
17	10	28	Na/Tl/Ca/Ce 48/3/48/1	100	3800	1000	96

In lamp nr. 17 the filling comprised additionally 0.25mg InI. The volume of the discharge vessel ranged from 2.1 mm<sup>3</sup> for lamp nr. 15 to 2.4 mm<sup>3</sup> for the other lamps. All lamps showed very stable color properties over the listed life time.



## CLAIMS:

1. A metal halide lamp comprising a discharge vessel surrounded by an outer envelope with clearance and having a ceramic wall which encloses a discharge space filled with a filling comprising an inert gas, such as xenon (Xe), and an ionizable salt, wherein in said discharge space two electrodes are arranged whose tips have a mutual interspacing so as  
5 to define a discharge path between them, characterized in that said ionizable salt comprises NaI, TII,  $\text{CaI}_2$  and X-iodide, wherein X is one or more elements selected from the group comprising rare earth metals.
2. Lamp according to claim 1, wherein X is one or more elements selected from  
10 the group comprising Sc, Y, La, Ce, Pr, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Nd.
3. Lamp according to claim 1 or 2, wherein X is one or more elements selected from the group comprising Ce, Pr, Nd
- 15 4. Lamp according to claim 1, 2 or 3, wherein the molar percentage ratio X-iodide/(NaI + TII +  $\text{CaI}_2$  + X-iodide) lies between 0 and 10%, in particular between 0,5 and 7%, more in particular between 1 and 6%.
5. Lamp according to claim 1, 2, 3 or 4, wherein the molar percentage ratio  
20  $\text{CaI}_2$ /(NaI + TII +  $\text{CaI}_2$  + X-iodide) lies between 20 and 95%.
6. Lamp according to any of the preceding claims 1 through 5, wherein the amount of NaI, TII,  $\text{CaI}_2$  and X-iodide lies between 0,001 and 0,5  $\text{g/cm}^3$ , in particular between 0,025 and 0,3  $\text{g/cm}^3$ .  
25
7. Lamp according to any of the preceding claims 1 through 6, emitting light during stable nominal operation having a color temperature  $T_c$  above 3500K, wherein the filling of the discharge space also comprises a halide selected from Mn and In.

8. Lamp according to any of the preceding claims 1 through 7, wherein the filling comprises Hg.

5 9. Metal halide lamp to be used in a vehicle headlamp according to any of the preceding claims 1 through 6.

10 10. Method for manufacturing a vehicle headlamp according to any of the preceding claims 1 through 6, wherein the vehicle headlamp is provided with a metal halide lamp comprising a discharge vessel surrounded by an outer envelope with clearance and having a ceramic wall which encloses a discharge space filled with a filling comprising an inert gas, such as xenon (Xe), and an ionizable salt, wherein in said discharge space two electrodes are arranged whose tips have a mutual interspacing so as to define a discharge path between them, characterized in that said ionizable salt comprises NaI, TlI, CaI<sub>2</sub> and X-iodide, wherein X is selected from the group comprising rare earth metals.

**ABSTRACT:**

A metal halide lamp comprising a discharge vessel surrounded by an outer envelope with clearance and having a ceramic wall which encloses a discharge space filled with a filling comprising an inert gas, such as xenon (Xe), and an ionizable salt, wherein in said discharge space two electrodes are arranged whose tips have a mutual interspacing so as to define a discharge path between them, with the special feature that said ionizable salt comprises NaI, TII, CaI<sub>2</sub> and X-iodide wherein X is selected from the group comprising rare earth metals.

Fig. 1





1/2

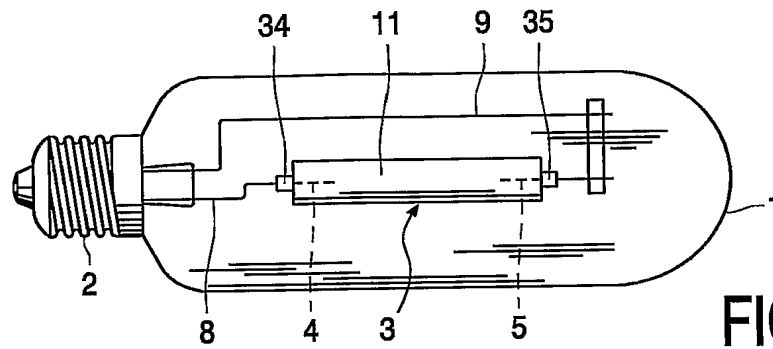


FIG. 1

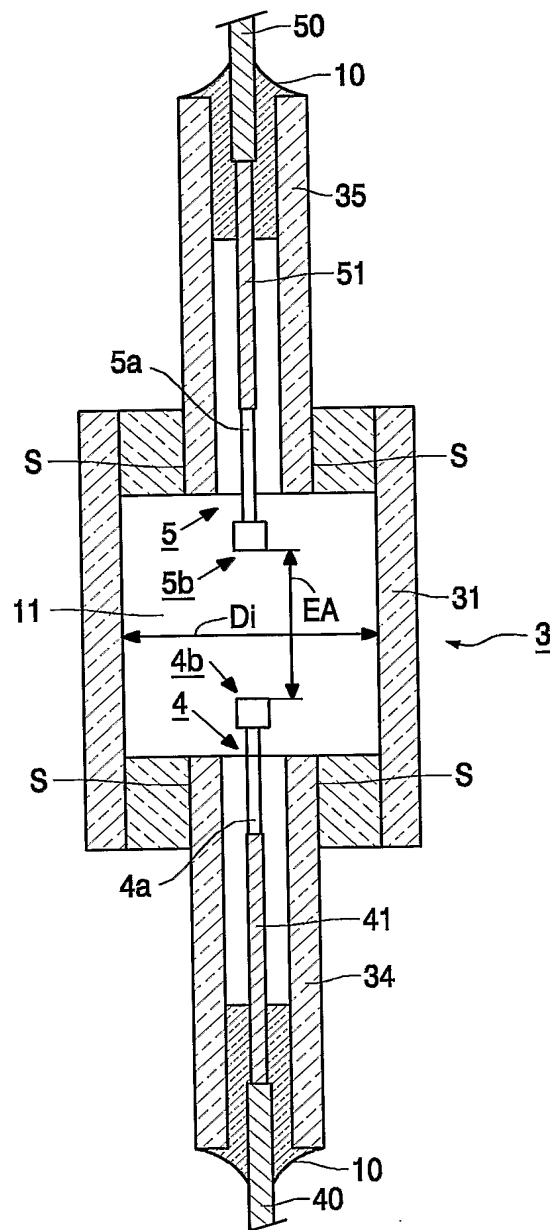


FIG. 2

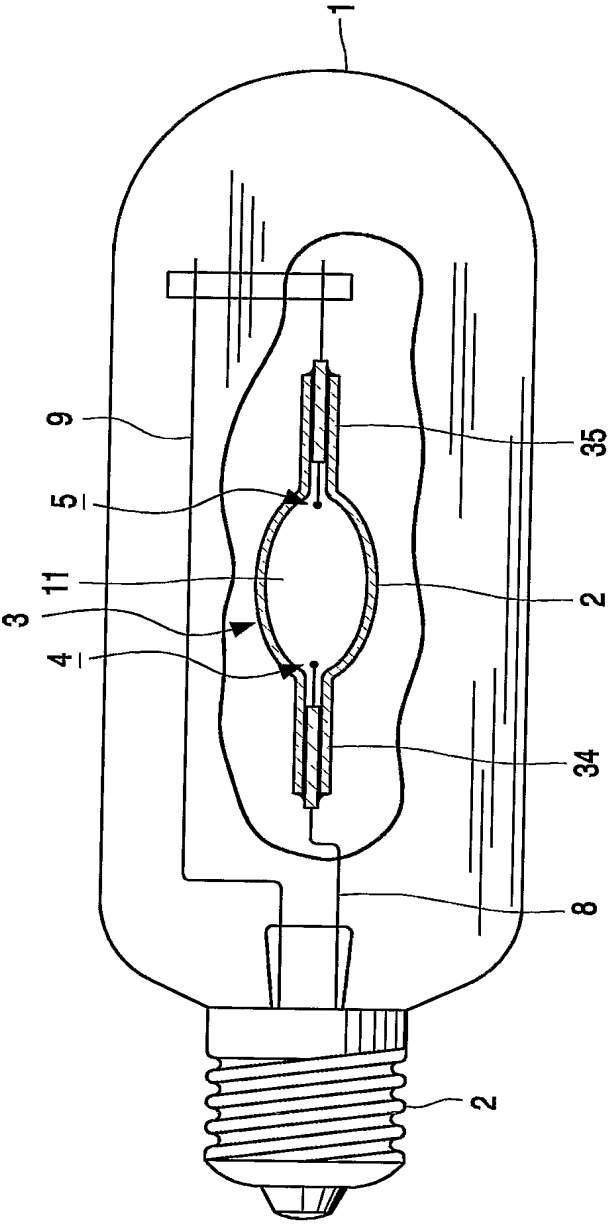


FIG. 3